

Liang-Zhong Xu,\* Shuang-Hua  
Yang, Zhi-Guo Jiang, Guan-Ping  
Yu and Guo-Dong SiCollege of Chemistry and Molecular  
Engineering, Qingdao University of Science  
and Technology, Qingdao 266042, People's  
Republic of China

Correspondence e-mail: qknhs@yahoo.com.cn

## Key indicators

Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$   
 $R$  factor = 0.043  
 $wR$  factor = 0.130  
Data-to-parameter ratio = 16.2For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.1-(2,4-Difluorophenyl)-1-[3-(2-fluorophenyl)-  
oxiran-2-yl]methanoneIn the title molecule,  $\text{C}_{15}\text{H}_9\text{F}_3\text{O}_2$ , the bond lengths and angles  
are within normal ranges. The crystal packing is stabilized by  
weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds.

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## Comment

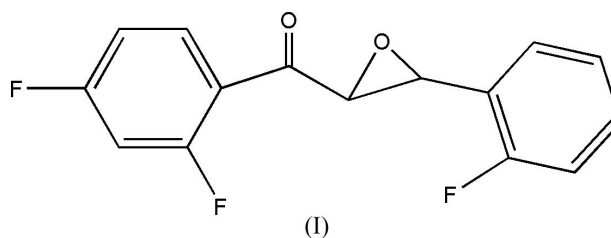
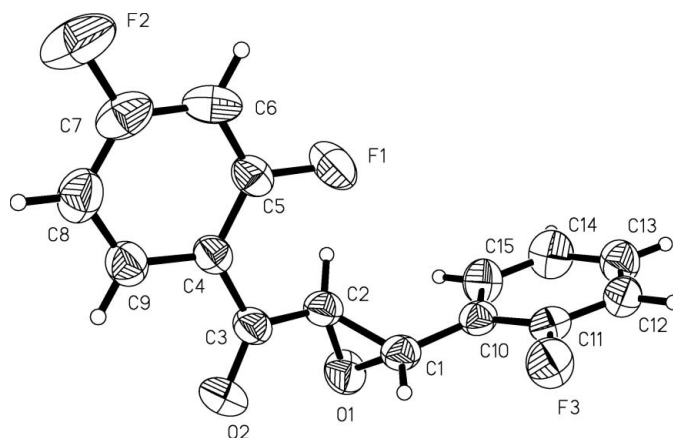
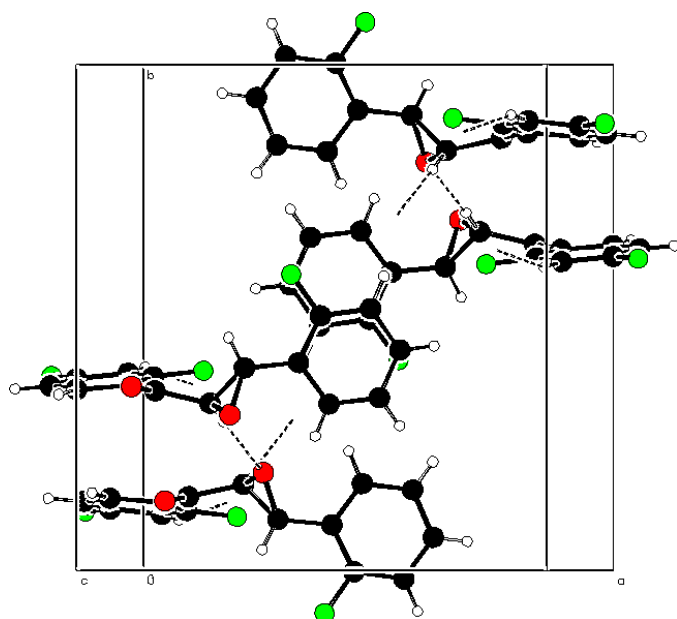
Epoxides are often used in the synthesis of organic  
compounds (Taylor *et al.*, 2000; Subburaj *et al.*, 2002). Their  
ring-opening reactions (Wroblewski & Halajewska, 2002) can  
be used in the preparation of polyfunctional cyclic compounds.  
They can also be applied in stereo- and regioselective syntheses  
(Tan *et al.*, 2004), which could afford several diastereo-  
meric epoxides (Subbiah *et al.*, 1999). In our search for new  
epoxides, the title compound, (I), has been synthesized. We  
report its crystal structure here.The bond lengths and angles in (I) (Fig. 1) are within  
normal ranges (Table 1). The geometric parameters of the  
oxirane ring, C1/C2/O1, are in good agreement with those  
found in the literature (Baures *et al.*, 1990). The mean planes  
 $P1$  (F1/F2/C3–C9) and  $P2$  (F3/C1/C10–C15) make dihedral  
angles of  $54.71(3)$  and  $77.79(2)^\circ$ , respectively, with the  
oxirane ring.

Figure 1

A view of (I), with the atom-numbering scheme. Displacement ellipsoids  
are drawn at the 40% probability level.



**Figure 2**  
The packing of (I), viewed approximately along the *c* axis. Intermolecular C—H...O hydrogen bonds are shown as dashed lines.

The crystal packing (Fig. 2) is stabilized by weak intermolecular C—H...O hydrogen bonds (Table 2).

## Experimental

A mixture of 2-bromo-1-(2,4-difluorophenyl)ethanone (0.02 mol, 4.70 g), 2-fluorobenzaldehyde (0.02 mol, 2.48 g) and water (50 ml) was stirred for 1 h at room temperature. The solution was then filtered and concentrated, and the solid product was purified by recrystallization from ethanol to afford the title compound (yield 4.25 g, 85%). Single crystals of (I) suitable for X-ray measurements were obtained by recrystallization from ethyl acetate at room temperature.

### Crystal data

$C_{15}H_9F_3O_2$	$D_x = 1.502 \text{ Mg m}^{-3}$
$M_r = 278.22$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 1564 reflections
$a = 12.429 (2) \text{ \AA}$	$\theta = 2.3\text{--}22.0^\circ$
$b = 13.359 (2) \text{ \AA}$	$\mu = 0.13 \text{ mm}^{-1}$
$c = 7.6176 (13) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 103.414 (2)^\circ$	Block, colourless
$V = 1230.3 (3) \text{ \AA}^3$	$0.24 \times 0.18 \times 0.16 \text{ mm}$
$Z = 4$	

### Data collection

Bruker SMART CCD area-detector diffractometer	2949 independent reflections
$\varphi$ and $\omega$ scans	1765 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.024$
$T_{\text{min}} = 0.962$ , $T_{\text{max}} = 0.980$	$\theta_{\text{max}} = 28.0^\circ$
8121 measured reflections	$h = -16 \rightarrow 16$
	$k = -10 \rightarrow 17$
	$l = -10 \rightarrow 9$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0455P)^2 + 0.1661P]$
$R[F^2 > 2\sigma(F^2)] = 0.043$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.130$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.05$	$\Delta\rho_{\text{max}} = 0.35 \text{ e \AA}^{-3}$
2949 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
182 parameters	Extinction correction: <i>SHELXL97</i> (Sheldrick, 1997)
H-atom parameters constrained	Extinction coefficient: 0.011 (2)

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

F1—C5	1.341 (2)	O1—C1	1.424 (2)
F2—C7	1.344 (2)	O2—C3	1.2124 (19)
F3—C11	1.358 (2)	C1—C2	1.474 (2)
O1—C2	1.418 (2)		
C2—O1—C1	62.47 (11)	O1—C2—C1	58.94 (11)
C2—C1—C10	122.48 (15)	C1—C2—C3	118.60 (16)

**Table 2**

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
C2—H2...O1 <sup>i</sup>	0.98	2.38	3.356 (2)	173
C6—H6...O2 <sup>ii</sup>	0.93	2.39	3.262 (2)	157

Symmetry codes: (i)  $x, -y + \frac{3}{2}, z + \frac{1}{2}$ ; (ii)  $x, y, z + 1$ .

All H atoms were placed in calculated positions, with C—H = 0.93 or 0.98  $\text{\AA}$ , and included in the final cycles of refinement using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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